

**Derivatives of Methyl 5-Methyl-4-oxo-1,2,4,5,8,8a-hexahydrocyclopropa[3]pyrrolo[3,2-e]indole-3-carboxylate: A Case of Inverse Electronic Effects on the Reactivity of CC-1065 Derivatives** [*J. Am. Chem. Soc.* **2001**, *123* (21), 5102–5103]. Luis Castedo, José Delamano, Juan Enjo, Jesús Fernández, Dolores G. Grávalos, Ramón Leis, Carmen López, Carlos F. Marcos, Ana Ríos, and Gabriel Tojo\*

Page 5102. After ref 5, a new reference (5a) must be included with the following text:

For previous uses of the photocyclization of stilbenoids in the preparations of CC-1065 analogues, see: (a) Rawal, V. H.; Cava, M. P. *J. Chem. Soc., Chem. Commun.* **1984**, 1526. (b) Rawal, V. H.; Jones, R. J.; Cava, M. P. *Tetrahedron Lett.* **1985**, *26*, 2423. (c) Jones, R. J.; Cava, M. P. *J. Chem. Soc., Chem. Commun.* **1986**, 826. (d) Rawal, V. H.; Cava, M. P. *J. Am. Chem. Soc.* **1986**, *108*, 2110. (e) Rawal, V. H.; Jones, R. J.; Cava, M. P. *J. Org. Chem.* **1987**, *52*, 19. (f) Drost, K. J.; Jones, R. J.; Cava, M. P. *J. Org. Chem.* **1989**, *54*, 5985. (g) Rajeswari, S.; Adesomoju, A. A.; Cava, M. P. *J. Heterocycl. Chem.* **1989**, *26*, 557. (h) Drost, K. J.; Cava, M. P. *J. Org. Chem.* **1991**, *56*, 2240.

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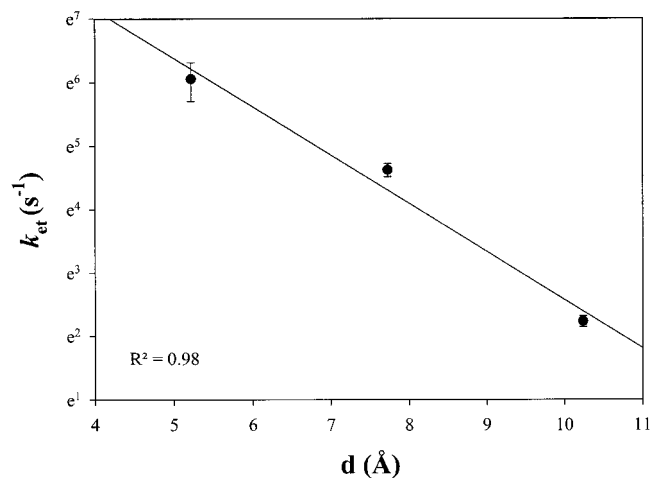
**Ion-Induced Rectification of Nanoparticle Quantized Capacitance Charging in Aqueous Solutions** [*J. Am. Chem. Soc.* **2001**, *123*, 10607]. Shaowei Chen\* and Renjun Pei

Page 10613. In the original analysis of the gold nanoparticle electron-transfer rate constant ( $k_{et}$ ) from AC voltammetric measurements (Figure 4), the calculations are incorrect. The corrected  $k_{et}$  data are listed in the new Table 3 along with the new Figure 5. Note that the tunneling constant ( $\beta$ ) is practically unchanged ( $0.8 \text{ \AA}^{-1}$ ) while the linearity between  $\ln(k_{et})$  and  $d$  is improved ( $R^2 = 0.98$ ). We regret the error.

**Table 3.** Electron-Transfer Rate Constants ( $k_{et}$ ) of MPCs with Varied Protecting Ligands and Charge States ( $z$ ) in Different Solution Media<sup>a</sup>

$k_{et}$ ( $s^{-1}$ )	$z$						average <sup>b</sup>
	-2	-1	+1	+2	+3	+4	
C4Au (CH <sub>2</sub> Cl <sub>2</sub> )	642.1	396.6	495.7	349	392.3	284.8	426.7 ± 126
C4Au (H <sub>2</sub> O)			23.7	48.8	65.0	59.6	49.3 ± 18.3
C6Au (CH <sub>2</sub> Cl <sub>2</sub> )	113.4	83.7	108.5	100.4	106.7	98.3	101.8 ± 10.4
C6Au (H <sub>2</sub> O)			15.2	11.7	18.2	22.9	17 ± 4.7
C8Au (CH <sub>2</sub> Cl <sub>2</sub> )		9.0	8.2	9.9	10.0		9.3 ± 0.8

<sup>a</sup> In H<sub>2</sub>O, 0.10 M NH<sub>4</sub>PF<sub>6</sub> was used as the supporting electrolyte, while in CH<sub>2</sub>Cl<sub>2</sub>, 0.10 M TBAP was used instead. <sup>b</sup> Averaged over all charge states.



**Figure 5.**

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